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FILE COVERS 1861 TO 1 FEB 2008 (20080201/ED)

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=> s (trigonal and huntite and borate).ti.

MISSING OPERATOR BORATE).TI.

The search profile that was entered contains terms or nested terms that are not separated by a logical operator.

=> s (trigonal and huntite and borate)/ti

37 TRIGONAL/TI

2 HUNTITE/TI

156 BORATE/TI

L1 0 (TRIGONAL AND HUNTITE AND BORATE)/TI

=> s (huntite)/ti

L2 2 (HUNTITE)/TI

=> d iall

L2 ANSWER 1 OF 2 DISSABS COPYRIGHT (C) 2008 ProQuest Information and

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ACCESSION NUMBER: 2006:33831 DISSABS Order Number: AAI3194090 TITLE: A new trigonal huntite material and subgroup

relationships between crystallographic space groups

AUTHOR: Hruschka, Michael A. [Ph.D.]; Keszler, Douglas A. [advisor]

CORPORATE SOURCE: Oregon State University (0172)

SOURCE: Dissertation Abstracts International, (2005) Vol. 66, No.

10B, p. 5394. Order No.: AAI3194090. 1172 pages.

ISBN: 0-542-37265-7.

DOCUMENT TYPE: Dissertation

FILE SEGMENT: DAI LANGUAGE: English

ENTRY DATE: Entered STN: 20060621

Last Updated on STN: 20060621

ABSTRACT: The use of an adjacency matrix to determine distant (not

maximal or minimal) subgroup and supergroup
relationships between crystallographic space-group types

is described. Full lists of space-group types that are supergroups and subgroups for every space-group type were compiled. A list of the space-group types connected to each space-group type by combined maximal

subgroup/minimal supergroup paths was compiled. Each of these lists was also compiled in matrix form, showing

for each pair of space-group types whether one is a

subgroup of the other and how many maximal subgroup, minimal supergroup, or combination of maximal subgroup and minimal supergroup steps are required to connect them. A method for using these lists and matrices to construct shortest path subgroup/supergroup graphs between space-group types was developed. From the matrices, statistics were compiled on the number of subgroup and supergroup paths of lengths one to six between space-group types, the average, median, and expected shortest path length between space-group types, and the number of space-group types each spacegroup type has as subgroups and supergroups. Correlations were sought between these properties and the number of organic and inorganic crystal structures of each space-group type. It was determined that organic compounds tend to crystallize in space-space groups that have many space-group types as supergroups and few space-group types as subgroups. The 17 most prevalent organic structure space-group types, comprising 90% of organic structures, were found to be closely related (paths of length 1 or 2) by subgroup/supergroup paths to each of two space-group types: P2 1 and P21/c. Other space-group type were found to be related to space-group types comprising more than 90% of organic structures by paths of length one or two. Properties of graphs and trees consisting exclusively of type I or type II subgroup relationships are discussed. The subgroup relationships work was motivated by the structure determination of a new trigonal huntite material, vttrium lanthanum scandium borate. Linear and nonlinear optical properties, the structure, and the composition range of this material are discussed.

CLASSIFICATION: 0488 CHEMISTRY, INORGANIC; 0794 ENGINEERING, MATERIALS

SCIENCE

=> d iall 2

L2 ANSWER 2 OF 2 DISSABS COPYRIGHT (C) 2008 ProQuest Information and

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ACCESSION NUMBER: 71:29107 DISSABS Order Number: AAR7214316

TITLE: A CALORIMETRIC DETERMINATION OF THE STABILITY, ENTROPY,

HEAT, AND GIBBS ENERGY OF FORMATION FOR THE CARBONATE

MINERALS HUNTITE, NESQUEHONITE, ARTINITE, AND

HYDROMAGNESITE

AUTHOR: HEMINGWAY, BRUCE SHERMAN [PH.D.]
CORPORATE SOURCE: UNIVERSITY OF MINNESOTA (0130)

SOURCE: Dissertation Abstracts International, (1971) Vol. 32, No.

11B, p. 6475. Order No.: AAR7214316. 278 pages.

DOCUMENT TYPE: Dissertation

FILE SEGMENT: DAI LANGUAGE: English

ENTRY DATE: Entered STN: 19921118

Last Updated on STN: 19921118

CLASSIFICATION: 0372 GEOLOGY

=>

Executing the logoff script...

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SESSION WILL BE HELD FOR 120 MINUTES
STN INTERNATIONAL SESSION SUSPENDED AT 18:18:28 ON 16 FEB 2008

* * * * * RECONNECTED TO STN INTERNATIONAL * * * * * * * SESSION RESUMED IN FILE 'DISSABS' AT 19:34:39 ON 16 FEB 2008 FILE 'DISSABS' ENTERED AT 19:34:39 ON 16 FEB 2008 COPYRIGHT (C) 2008 ProQuest Information and Learning Company; All Rights Reserved.

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Property values tagged with IC are from the ZIC/VINITI data file provided by InfoChem.

STRUCTURE FILE UPDATES: 15 FEB 2008 HIGHEST RN 1003765-97-6 DICTIONARY FILE UPDATES: 15 FEB 2008 HIGHEST RN 1003765-97-6

New CAS Information Use Policies, enter HELP USAGETERMS for details.

TSCA INFORMATION NOW CURRENT THROUGH June 29, 2007

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REGISTRY includes numerically searchable data for experimental and predicted properties as well as tags indicating availability of experimental property data in the original document. For information on property searching in REGISTRY, refer to:

http://www.cas.org/support/stngen/stndoc/properties.html

=> d 1-5

- L5 ANSWER 1 OF 5 REGISTRY COPYRIGHT 2008 ACS on STN
- RN 910055-88-8 REGISTRY
- ED Entered STN: 10 Oct 2006
- CN Scandium ytterbium oxide (ScYbO3) (9CI) (CA INDEX NAME)
- MF O . Sc . Yb
- AF 03 Sc Yb
- CI TIS
- SR CA
- LC STN Files: CA, CAPLUS

Component		Ratio	 	Component Registry Number
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0		3		17778-80-2
Yb		1		7440-64-4
Sc		1		7440-20-2

PROPERTY DATA AVAILABLE IN THE 'PROP' FORMAT

- 1 REFERENCES IN FILE CA (1907 TO DATE)
- 1 REFERENCES IN FILE CAPLUS (1907 TO DATE)
- L5 ANSWER 2 OF 5 REGISTRY COPYRIGHT 2008 ACS on STN
- RN 73146-02-8 REGISTRY
- ED Entered STN: 16 Nov 1984
- CN Scandium terbium oxide (ScTbO3) (CA INDEX NAME)

OTHER NAMES:

- CN Terbium scandate (TbScO3)
- MF O . Sc . Tb
- AF 03 Sc Tb
- CI TIS
- LC STN Files: CA, CAPLUS, USPATFULL

Component	 +	Ratio	Component Registry Number
	+		
0	- 1	3	17778-80-2
Tb	1	1	7440-27-9
Sc	1	1	7440-20-2

PROPERTY DATA AVAILABLE IN THE 'PROP' FORMAT

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- 1 REFERENCES TO NON-SPECIFIC DERIVATIVES IN FILE CA
- 8 REFERENCES IN FILE CAPLUS (1907 TO DATE)
- L5 ANSWER 3 OF 5 REGISTRY COPYRIGHT 2008 ACS on STN
- RN 25962-01-0 REGISTRY
- ED Entered STN: 16 Nov 1984
- CN Antimony scandium oxide (SbScO4) (CA INDEX NAME)

OTHER CA INDEX NAMES:

- CN Antimonic acid (H3SbO4), scandium(3+) salt (1:1) (8CI)
- CN Scandium antimonate(V) (ScSbO4) (7CI)

OTHER NAMES:

- CN Scandium antimonate (ScSbO4)
- DR 61419-99-6
- MF O . Sb . Sc
- AF 04 Sb Sc
- CI TIS
- LC STN Files: CA, CAOLD, CAPLUS, IFICDB, IFIPAT, IFIUDB, USPAT2, USPATFULL, USPATOLD

Component		Ratio	1	Component
	 		l	Registry Number
	+		+-	
0		4		17778-80-2
Sb		1		7440-36-0
Sc	1	1	1	7440-20-2

- 7 REFERENCES IN FILE CA (1907 TO DATE)
- 7 REFERENCES IN FILE CAPLUS (1907 TO DATE)
- 1 REFERENCES IN FILE CAOLD (PRIOR TO 1967)
- L5 ANSWER 4 OF 5 REGISTRY COPYRIGHT 2008 ACS on STN

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RN
  12351-60-9 REGISTRY
ED Entered STN: 16 Nov 1984
CN Niobium scandium oxide (NbScO4) (CA INDEX NAME)
OTHER CA INDEX NAMES:
CN Scandium niobate(V) (6CI, 7CI)
OTHER NAMES:
CN Scandium niobate (ScNbO4)
DR 12533-67-4
MF Nb.O.Sc
AF Nb O4 Sc
CI COM, TIS
LC STN Files: CA, CAOLD, CAPLUS, CSCHEM, USPATFULL
 Component | Ratio | Component | Registry Number
0 | 4 | 17778-80-2
          1 1 7440-20-2
1 1 7440-03-1
Sc
Nb
**PROPERTY DATA AVAILABLE IN THE 'PROP' FORMAT**
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            29 REFERENCES IN FILE CAPLUS (1907 TO DATE)
             5 REFERENCES IN FILE CAOLD (PRIOR TO 1967)
    ANSWER 5 OF 5 REGISTRY COPYRIGHT 2008 ACS on STN
L5
RN 135-58-0 REGISTRY
ED Entered STN: 16 Nov 1984
CN Thianthrene, 2,7-dimethyl- (CA INDEX NAME)
OTHER NAMES:
CN 2,7-Dimethylthianthrene
CN Cutilen
CN Cutosolo
CN Mesulfen
CN Mesulphen
CN Mitabol
CN Mitigal
CN Neosulfine
CN Odylen
CN Peligal
CN Scabol
CN Sudermo
CN Thianthol
CN
   Thianthrol
MF
   C14 H12 S2
CI COM
LC
   STN Files: AGRICOLA, ANABSTR, BEILSTEIN*, BIOSIS, CA, CAOLD, CAPLUS,
      CASREACT, CHEMCATS, CHEMLIST, DDFU, DRUGU, EMBASE, MEDLINE, MRCK*,
      PROMT, PS, RTECS*, TOXCENTER, USAN, USPATFULL, USPATOLD
        (*File contains numerically searchable property data)
    Other Sources: EINECS**, WHO
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(**Enter CHEMLIST File for up-to-date regulatory information)

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Entered STN: 12 Feb 2004

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               6 REFERENCES IN FILE CAOLD (PRIOR TO 1967)
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L1
L2
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L3
              0 S LA!!!Y!!!SC!BO?
L4
              0 S LA(3W)Y(3W)SC!BO?
L5
              5 S SC!BO?
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L2
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              0 S LA!!!Y!!!SC!BO?
L3
L4
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             5 S SC!BO?
L6
             62 S LANTHANUM (4A) SCANDIUM (4A) BORATE
L7
             14 S LANTHANUM (4A) YTTRIUM (4A) SCANDIUM (4A) BORATE
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=> s 17
L8
            24 L7
=> s optical
L9 884666 OPTICAL
=> s 18 and 19
    6 L8 AND L9
L10
=> d 6 all
L10 ANSWER 6 OF 6 CA COPYRIGHT 2008 ACS on STN
     140:119647 CA Full-text
ΑN
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Rare earth scandoborate-based nonlinear optical materials and

UV-emitting sources employing the materials ΙN Keszler, Douglas A.; Stone-Sundberg, Jennifer L.; Ye, Ning; Hruschka, Michael A. The State of Oregon Acting by and Through the State Board of Higher PAEducation, On Behalf of Oregon State University, USA PCT Int. Appl., 29 pp. SO CODEN: PIXXD2 Patent DT English LA ΙC ICM C01B 73-10 (Optical, Electron, and Mass Spectroscopy and Other Related CC Properties) FAN.CNT 1 KIND DATE APPLICATION NO. DATE PATENT NO. ____ _____ ______ _____ WO 2004007352 A2 20040122 WO 2003-US22075 20030714 WO 2004007352 A3 20040902 B1 20050224 WO 2004007352 W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG AU 2003249258 A1 20040202 AU 2003-249258 20030714 US 2005247918 A1 20051110 US 2005-520364 20050103 Р PRAI US 2002-395681P 20020712 W WO 2003-US22075 20030714 CLASS PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES _____ WO 2004007352 ICM C01B IPCI C01B [ICM, 7] IPCR C01B0035-00 [I,C*]; C01B0035-12 [I,A]; C30B0009-00 [I,A]; C30B0009-00 [I,C*]; G02F0001-35 [I,C*]; G02F0001-355 [I,A] ECLA C01B035/12; C30B009/00+29/10; C30B009/00+29/22; G02F001/355C AU 2003249258 IPCI G02B0005-20 [ICM, 7]; G02F0001-35 [ICS, 7]; C01F0003-00 [ICS, 7] IPCR C01B0035-00 [I,C*]; C01B0035-12 [I,A]; C30B0009-00 [I,A]; C30B0009-00 [I,C*]; G02F0001-35 [I,C*]; G02F0001-355 [I,A] US 2005247918 IPCI G02B0005-30 [ICM, 7] IPCR G02B0005-30 [I,C*]; G02B0005-30 [I,A]; G02F0001-35 [I,C*]; G02F0001-355 [I,A] NCL 252/585.000 G02F001/355C ECLA Nonlinear optical materials are described having the general formula AΒ MxM'yScz(BO3)4 where M and M' are selected from La, Pr, Sm, Eu, Gd, Tb, Dy, Ho, Er, Tm, Yb, Lu, Sc and Y; and the sum of x, y, and z is \approx 4. One example of such a material is La0.7Y0.3Sc3(BO3)4. Exemplary crystalline materials according to the general formula exhibit useful optical characteristics and desirable phys. properties for nonlinear optical applications. Compns. and UV devices using the nonlinear optical materials are also described.

scandoborate lanthanum nonlinear optical crystal UV source NLO;

ST

```
rare earth scandium borate nonlinear optical crystal UV source
ΙT
     Nonlinear optical materials
     UV sources
        (rare earth scandoborate-based nonlinear optical materials
        and UV-emitting sources employing materials)
ΤТ
     RL: DEV (Device component use); TEM (Technical or engineered material
     use); USES (Uses)
        (rare earth scandoborate-based nonlinear optical materials
        and UV-emitting sources employing materials)
ΤТ
     Rare earth compounds
     RL: DEV (Device component use); TEM (Technical or engineered material
     use); USES (Uses)
        (scandoborate; rare earth scandoborate-based nonlinear optical
        materials and UV-emitting sources employing materials)
ΙT
     1314-36-9, Yttrium oxide, uses 13453-69-5, Lithium borate
     RL: NUU (Other use, unclassified); USES (Uses)
        (flux; rare earth scandoborate-based nonlinear optical
        materials prepared using)
     554-13-2, Lithium carbonate 1303-86-2, Boron oxide B203, uses
ΤТ
     RL: NUU (Other use, unclassified); RCT (Reactant); RACT (Reactant or
     reagent); USES (Uses)
        (flux; rare earth scandoborate-based nonlinear optical
        materials prepared using)
     648431-00-9P, Lanthanum scandium yttrium borate
ΤT
     (La0.7Sc3Y0.3(BO3)4)
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        and UV-emitting sources employing materials)
     7429-91-6D, Dysprosium, compds. 7439-91-0D, Lanthanum, compds.
ΤТ
     7439-94-3D, Lutetium, compds. 7440-10-0D, Praseodymium, compds. 7440-19-9D, Samarium, compds. 7440-20-2D, Scandium, compds. 7440-27-9D, Terbium, compds. 7440-30-4D, Thulium, compds. 7440-52-0D,
     Erbium, compds. 7440-53-1D, Europium, compds. 7440-54-2D, Gadolinium,
     compds. 7440-60-0D, Holmium, compds. 7440-64-4D, Ytterbium, compds.
     7440-65-5D, Yttrium, compds.
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     648431-04-3, Lanthanum scandium yttrium borate
     (La0.75Sc3Y0.25(BO3)4)
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        (rare earth scandoborate-based nonlinear optical materials
        and UV-emitting sources employing materials)
=> d all 5
L10 ANSWER 5 OF 6 CA COPYRIGHT 2008 ACS on STN
ΑN
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ED
     Entered STN: 22 Apr 2004
     Phosphor blends and backlight sources for color liquid crystal displays
TΙ
     Setlur, Anant Achyut; Srivastava, Alok Mani; Comanzo, Holly Ann
IN
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PA

General Electric Company, USA

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SO
    U.S. Pat. Appl. Publ., 11 pp.
    CODEN: USXXCO
DT
    Patent
LA
    English
IC
    ICM C09K011-08
INCL 349069000; 252301400R; 252301400P; 252301400H; 252301400F; 252301600F;
    252301600P; 252301400S
    73-5 (Optical, Electron, and Mass Spectroscopy and Other Related
CC
    Properties)
    Section cross-reference(s): 74
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    PATENT NO.
                       KIND
                              DATE
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    US 6809781
                       B2 20041026
    TW 282883
                       B 20070621 TW 2003-92125345
                                                                20030915

      JP 2004168996
      A
      20040617
      JP 2003-329248

      EP 1403355
      A1
      20040331
      EP 2003-255943

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                       G02F0001-1335 [ICS,7]; G02F0001-13 [ICS,7,C*];
                       H01J0061-44 [ICS, 7]; H01J0061-38 [ICS, 7, C*];
                       H05B0033-12 [ICS,7]; H05B0033-14 [ICS,7]; F21Y0101-02
                       [ICS, 7]; F21Y0105-00 [ICS, 7]
                IPCR
                       C09K0011-08 [I,A]; C09K0011-08 [I,C*]; C09K0011-57
                       [I,A]; C09K0011-57 [I,C*]; C09K0011-77 [I,A];
                       C09K0011-77 [I,C*]; G02F0001-13 [I,C*]; G02F0001-13357
                       [I,A]; H05B0033-14 [I,A]; H05B0033-14 [I,C*]
                FTERM 2H091/FA02Y; 2H091/FA08X; 2H091/FA08Z; 2H091/FA14Z;
                       2H091/FA23Z; 2H091/FA31Z; 2H091/FA42Z; 2H091/FA44Z;
                       2H091/FA45Z; 2H091/FB02; 2H091/FB06; 2H091/FB12;
                       2H091/FB13; 2H091/FC01; 2H091/FC02; 2H091/FD06;
                       2H091/FD11; 2H091/FD22; 2H091/HA06; 2H091/LA15;
                       2H091/LA30; 3K007/AB04; 3K007/BB06; 3K007/DB03;
                       4H001/CA04; 4H001/CA05; 4H001/XA01; 4H001/XA05;
                       4H001/XA08; 4H001/XA09; 4H001/XA12; 4H001/XA13;
                       4H001/XA14; 4H001/XA15; 4H001/XA16; 4H001/XA17;
                       4H001/XA20; 4H001/XA21; 4H001/XA30; 4H001/XA31;
                       4H001/XA32; 4H001/XA35; 4H001/XA38; 4H001/XA39;
                       4H001/XA49; 4H001/XA56; 4H001/XA57; 4H001/XA59;
                       4H001/XA62; 4H001/XA64; 4H001/XA65; 4H001/XA71;
                       4H001/YA25; 4H001/YA58; 4H001/YA63; 4H001/YA65;
                       4H001/YA83
EP 1403355
                       C09K0011-08 [ICM, 7]; C09K0011-77 [ICS, 7]; G02F0001-1335
                IPCI
                       [ICS, 7]; G02F0001-13 [ICS, 7, C*]; H01J0061-00 [ICS, 7];
                       H01L0033-00 [ICS,7]
                IPCR
                       C09K0011-08 [I,C*]; C09K0011-08 [I,A]; C09K0011-56
                       [I,C*]; C09K0011-56 [I,A]; C09K0011-57 [I,C*];
                       C09K0011-57 [I,A]; C09K0011-59 [I,C*]; C09K0011-59
                       [I,A]; C09K0011-61 [I,C*]; C09K0011-61 [I,A];
                       C09K0011-64 [I,C*]; C09K0011-64 [I,A]; C09K0011-66
                       [I,C*]; C09K0011-66 [I,A]; C09K0011-70 [I,C*];
                       C09K0011-71 [I,A]; C09K0011-73 [I,A]; C09K0011-77
                       [I,C*]; C09K0011-77 [I,A]; C09K0011-78 [I,A];
                       C09K0011-80 [I,A]; C09K0011-82 [I,A]; C09K0011-84
                       [I,A]; F21S0002-00 [I,C*]; F21S0002-00 [I,A];
                       F21Y0101-02 [N,A]; F21Y0105-00 [N,A]; G02F0001-13
                       [I,C*]; G02F0001-13357 [I,A]; H01J0061-38 [I,C*];
                       H01J0061-44 [I,A]; H01L0051-50 [I,C*]; H01L0051-50
                       [I,A]; H05B0033-12 [I,C*]; H05B0033-12 [I,A];
                       H05B0033-14 [I,C*]; H05B0033-14 [I,A]
                ECLA
                       C09K011/08E; C09K011/57; C09K011/77N6; C09K011/77N10B;
                       C09K011/77N10B2; C09K011/77N12; C09K011/77S12;
                       G02F001/13357L; H05B033/14
KR 2004026628
                       C09K0011-78 [ICM, 7]; C09K0011-77 [ICM, 7, C*]
                IPCI
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ECLA
                        C09K011/08E; C09K011/57; C09K011/77N6; C09K011/77N10B;
                        C09K011/77N10B2; C09K011/77N12; C09K011/77S12;
                        G02F001/13357L; H05B033/14
CN 1495486
                 IPCI
                        G02F0001-1335 [ICM,7]; G02F0001-1335 [ICS,7];
                        G02F0001-13 [ICS,7,C*]; C09K0011-00 [ICS,7]
                        C09K0011-08 [I,C*]; C09K0011-08 [I,A]; C09K0011-56
                 IPCR
                        [I,C*]; C09K0011-56 [I,A]; C09K0011-57 [I,C*];
                        C09K0011-57 [I,A]; C09K0011-59 [I,C*]; C09K0011-59
                        [I,A]; C09K0011-61 [I,C*]; C09K0011-61 [I,A];
                        C09K0011-64 [I,C*]; C09K0011-64 [I,A]; C09K0011-66
                        [I,C*]; C09K0011-66 [I,A]; C09K0011-70 [I,C*];
                        C09K0011-71 [I,A]; C09K0011-73 [I,A]; C09K0011-77
                        [I,C*]; C09K0011-77 [I,A]; C09K0011-78 [I,A];
                        C09K0011-80 [I,A]; C09K0011-82 [I,A]; C09K0011-84
                        [I,A]; F21S0002-00 [I,C*]; F21S0002-00 [I,A];
                        F21Y0101-02 [N,A]; F21Y0105-00 [N,A]; G02F0001-13
                        [I,C*]; G02F0001-13357 [I,A]; H01J0061-38 [I,C*];
                        H01J0061-44 [I,A]; H01L0051-50 [I,C*]; H01L0051-50
                        [I,A]; H05B0033-12 [I,C*]; H05B0033-12 [I,A];
                        H05B0033-14 [I,C*]; H05B0033-14 [I,A]
                 ECLA
                        C09K011/08E; C09K011/57; C09K011/77N6; C09K011/77N10B;
                        C09K011/77N10B2; C09K011/77N12; C09K011/77S12;
                        G02F001/13357L; H05B033/14
     Phosphor compns. which comprises at least one phosphor emitting in each of the
AΒ
     blue, green, and red regions of the visible spectrum are described for use in
     a backlight source of a color liquid crystal display. Liquid crystal displays
     are described which include a backlighting system comprising a backlight
     source emitting light having a first spectrum at least in a range from \approx 300 -
     450 nm; and the above phosphor composition disposed to absorb light of at
     least a portion of the first spectrum and to emit light having a second
     spectrum different from the first spectrum; and a liquid crystal material
     disposed to receive light having the second spectrum.
ST
     phosphor blend backlight source color liq crystal display
ΙT
     Light sources
        (backlight; phosphor blends and backlight sources for liquid crystal
        displays)
ΙT
     Phosphors
        (blends; phosphor blends and backlight sources for liquid crystal
        displays)
ΙT
     Phosphors
        (blue-emitting; phosphor blends and backlight sources for liquid crystal
        displays)
     Liquid crystal displays
ΙT
        (color; phosphor blends and backlight sources for liquid crystal
        displays)
     Polysiloxanes, uses
ΤТ
     RL: DEV (Device component use); USES (Uses)
        (epoxy, phosphor dispersed in; phosphor blends and backlight sources
        for liquid crystal displays)
ΙT
     Phosphors
        (green-emitting; phosphor blends and backlight sources for liquid crystal
       displays)
ΙT
     Optical materials
        (light-scattering particles dispersed in polymer; phosphor blends and
        backlight sources for liquid crystal displays)
ΙT
     Acrylic polymers, uses
     Epoxy resins, uses
     Polysiloxanes, uses
     RL: DEV (Device component use); USES (Uses)
        (phosphor dispersed in; phosphor blends and backlight sources for liquid
```

crystal displays)

IT Transparent materials

(polymers, phosphor dispersed in; phosphor blends and backlight sources for liquid crystal displays)

IT Epoxy resins, uses

RL: DEV (Device component use); USES (Uses)

(polysiloxane-, phosphor dispersed in; phosphor blends and backlight sources for liquid crystal displays)

IT Phosphors

(red-emitting; phosphor blends and backlight sources for liquid crystal displays)

IT Electroluminescent devices

(semiconductor or organic, backlight source; phosphor blends and backlight sources for liquid crystal displays)

IT 675819-83-7

RL: DEV (Device component use); TEM (Technical or engineered material use); USES (Uses)

(Ce, Tb-codoped; phosphor blends and backlight sources for liquid crystal displays)

IT 12525-03-0, Calcium lanthanum sulfide (CaLa2S4)

RL: DEV (Device component use); TEM (Technical or engineered material use); USES (Uses)

(Ce-doped; phosphor blends and backlight sources for liquid crystal displays)

IT 173525-28-5, Gadolinium lanthanum lutetium yttrium oxide sulfide (Gd,La,Lu,Y)202S 675819-90-6 675819-91-7

RL: DEV (Device component use); TEM (Technical or engineered material use); USES (Uses)

(Eu, Bi-codoped; phosphor blends and backlight sources for liquid crystal displays)

IT 675819-89-3

RL: DEV (Device component use); PRP (Properties); TEM (Technical or engineered material use); USES (Uses)

(Eu,Mn-codoped; phosphor blends and backlight sources for liquid crystal displays)

IT 675819-88-2 675819-92-8

RL: DEV (Device component use); TEM (Technical or engineered material use); USES (Uses)

(Eu,Mn-codoped; phosphor blends and backlight sources for liquid crystal displays)

IT 1314-96-1, Strontium sulfide (SrS) 12535-38-5, Strontium yttrium sulfide (SrY2S4) 82992-94-7, Calcium strontium sulfide ((Ca,Sr)S)

RL: DEV (Device component use); TEM (Technical or engineered material use); USES (Uses)

(Eu-doped; phosphor blends and backlight sources for liquid crystal displays)

IT 12159-91-0, Germanium magnesium fluoride oxide (GeMg4F05.5)

RL: DEV (Device component use); PRP (Properties); TEM (Technical or engineered material use); USES (Uses)

IT 675819-87-1

RL: DEV (Device component use); TEM (Technical or engineered material use); USES (Uses)

(cerium-doped; phosphor blends and backlight sources for liquid crystal displays)

IT 7439-96-5, Manganese, uses 7440-27-9, Terbium, uses 7440-45-1, Cerium, uses 7440-53-1, Europium, uses 7440-69-9, Bismuth, uses 16397-91-4, Manganese(2+), uses 16910-54-6, Europium(2+), uses 18923-26-7, Cerium(3+), uses 19768-33-3, Manganese(4+), uses 22541-18-0,

Europium(3+), uses 22541-20-4, Terbium(3+), uses 23713-46-4, Bismuth(3+), uses

RL: DEV (Device component use); MOA (Modifier or additive use); TEM (Technical or engineered material use); USES (Uses)

(dopant; phosphor blends and backlight sources for liquid crystal displays)

IT 675819-79-1

RL: DEV (Device component use); PRP (Properties); TEM (Technical or engineered material use); USES (Uses)

(doped; phosphor blends and backlight sources for liquid crystal displays)

IT 473908-57-5

RL: DEV (Device component use); TEM (Technical or engineered material use); USES (Uses)

(doped; phosphor blends and backlight sources for liquid crystal displays)

ΙT 20775-37-5, Barium magnesium silicate (Ba3MgSi2O8) 76125-60-5, Aluminum strontium oxide (Al14Sr4O25) 97358-83-3, Aluminum barium oxide (Al8BaO13) 144920-98-9, Strontium borate metaphosphate oxide (Sr2(BO3)0.32(PO3)1.6800.68) 675819-80-4, Boron calcium strontium oxide phosphate (B0-2(Ca,Sr)1000-3(PO4)6) 675819-81-5, Strontium chloride oxide silicate (Sr4Cl4O0.5(Si2O5)1.5) 675819-82-6, Aluminum barium calcium strontium oxide (Al2(Ba,Ca,Sr)O4) 675819-84-8, Barium calcium strontium silicate ((Ba,Ca,Sr)2(SiO4)) 675819-85-9 675819-86-0 RL: DEV (Device component use); TEM (Technical or engineered material use); USES (Uses)

(europium-doped; phosphor blends and backlight sources for liquid crystal displays)

IT 675819-78-0

RL: DEV (Device component use); PRP (Properties); TEM (Technical or engineered material use); USES (Uses)

(phosphor blends and backlight sources for liquid crystal displays) RE.CNT 27 THERE ARE 27 CITED REFERENCES AVAILABLE FOR THIS RECORD RE

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Second-harmonic generation

Solubility
Space groups

Second-order nonlinear optical susceptibility

=> d 1-4 allL10 ANSWER 1 OF 6 CA COPYRIGHT 2008 ACS on STN AN 145:324196 CA Full-text ΕD Entered STN: 05 Oct 2006 Growth of nonlinear optical crystal Y0.57La0.72Sc2.71(BO3)4 ΤI Ye, Ning; Zhang, Yang; Chen, Wei; Keszler, Douglas A.; Aka, Gerard ΑU CS Fujian Institute of Research on the Structure of Matter, National Engineering Research Center for Optoelectronic Crystalline Materials, Chinese Academy of Sciences, Fuzhou, Fujian, 350002, Peop. Rep. China SO Journal of Crystal Growth (2006), 292(2), 464-467 CODEN: JCRGAE; ISSN: 0022-0248 PΒ Elsevier B.V. DT Journal LA English 73-10 (Optical, Electron, and Mass Spectroscopy and Other Related CC Properties) Section cross-reference(s): 65, 75 Large single crystals of Y0.57La0.72Sc2.71(BO3)4 were grown by a top-seeded AΒ high-temperature solution method. The high-energy optical absorption edge for polished pieces is at a wavelength of 190 nm. Sellmeier equations for the dispersion in the refractive indexes were determined from curve fitting of data obtained by the method of min. deviation. From modeling and optical measurements on powders, the nonlinear optical coefficient d11 is 1.35 pm/V. lanthanum scandium yttrium borate crystal growth nonlinear optical ST susceptibility; second harmonic generation lanthanum scandium yttrium borate; soly lanthanum scandium yttrium borate; melting point lanthanum scandium yttrium borate; birefringence point lanthanum scandium yttrium borate; space group lanthanum scandium yttrium borate; transmission optical lanthanum scandium yttrium borate; Moh hardness lanthanum scandium yttrium borate; XRD lanthanum scandium yttrium borate; refractive index lanthanum scandium yttrium borate; dispersion refractive index lanthanum scandium yttrium borate; absorption optical edge lanthanum scandium yttrium borate; density lanthanum scandium yttrium borate; mol wt lanthanum scandium yttrium borate Optical transmission IT(IR; of nonlinear optical crystal Y0.57La0.72Sc2.71(BO3)4) ΙT Hardness (mechanical) (Moh's; of nonlinear optical crystal Y0.57La0.72Sc2.71(BO3)4) ΙT Crystal growth (growth of nonlinear optical crystal Y0.57La0.72Sc2.71(BO3)4) ΙT Heat treatment (growth of nonlinear optical crystal Y0.57La0.72Sc2.71(BO3)4 with) ΙT Birefringence Density IR spectra Melting point Molecular weight Optical absorption edge Optical dispersion Optical transmission Refractive index

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X-ray diffraction
       (of nonlinear optical crystal Y0.57La0.72Sc2.71(BO3)4)
ΙT
    853030-11-2P, Lanthanum scandium yttrium borate
    (La0.72Sc2.71Y0.57(BO3)4)
    RL: PEP (Physical, engineering or chemical process); PNU (Preparation,
    unclassified); PRP (Properties); PYP (Physical process); PREP
    (Preparation); PROC (Process)
       (growth of nonlinear optical crystal Y0.57La0.72Sc2.71(BO3)4)
ΙT
    12664-58-3, Lithium borate li6b4o9
    RL: NUU (Other use, unclassified); USES (Uses)
       (growth of nonlinear optical crystal Y0.57La0.72Sc2.71(BO3)4
       usina)
    1303-86-2, Boron sesquioxide, processes 1312-81-8, Lanthanum sesquioxide
ΙT
    1314-36-9, Yttria, processes 12060-08-1, Scandia
    RL: CPS (Chemical process); PEP (Physical, engineering or chemical
    process); PROC (Process)
       (nonlinear optical crystal Y0.57La0.72Sc2.71(BO3)4 prepared
       using)
RE.CNT 12 THERE ARE 12 CITED REFERENCES AVAILABLE FOR THIS RECORD
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L10 ANSWER 2 OF 6 CA COPYRIGHT 2008 ACS on STN
AN 145:133778 CA Full-text
ED Entered STN: 03 Aug 2006
TI Quantum-splitting fluoride-based phosphors and radiation sources and
    displays incorporating same
ΙN
    Manivannan, Venkatesan; Srivastava, Alok Mani; Comanzo, Holly Ann
PA General Electric Company, USA
SO U.S. Pat. Appl. Publ., 22 pp.
    CODEN: USXXCO
DT Patent
LA English
INCL 252301400H; 252301400P; 252301400R; 252301500
    73-5 (Optical, Electron, and Mass Spectroscopy and Other Related
    Properties)
    Section cross-reference(s): 74
FAN.CNT 1
    PATENT NO.
                      KIND DATE APPLICATION NO. DATE
    _____
                      ____
                              _____
                                         ______
    US 2006151747 A1 20060713 US 2005-32910 20050110 US 7270773 B2 20070918
PRAI US 2005-32910
                             20050110
CLASS
PATENT NO.
              CLASS PATENT FAMILY CLASSIFICATION CODES
______
US 2006151747 INCL 252301400H
               IPCI C09K0011-77 [I,A]; C09K0011-85 [I,A]
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IPCR
                   C09K0011-77 [I,A]; C09K0011-77 [I,C]
            NCL
                   252/301.40H; 252/301.40P; 252/301.40R; 252/301.500;
                   252/301.4H0; 313/467.000; 313/468.000; 313/486.000;
                   313/487.000
            ECLA
                   C09K011/77N4B; C09K011/77P10; C09K011/77T2H;
                   C09K011/77T4B; H01J001/63
Phosphors are described by the general formula AGdF4:RE (A = K, Rb, and/or Cs;
and RE = rare earth metal activator(s) other than Gd). The phosphors may
comprise addnl. alkali metals and metals selected from V, Nb, W, Zr, Hf, Sb,
Ge, Sn, Bi, Ga, Zn, In, Cu, Ag, Er, Tm, and/or Pr. Phosphor blends
incorporating the phosphors are also described, as are light sources and
cathodoluminescent displays. Preparation of the phosphors using a solid-state
method without using HF is discussed.
quantum splitting alkali metal gadolinium fluoride phosphor; display
quantum splitting alkali metal gadolinium fluoride phosphor; light source
quantum splitting alkali metal gadolinium fluoride phosphor
Fluorides, uses
RL: DEV (Device component use); USES (Uses)
   (alkali metal gadolinium; quantum-splitting fluoride-based phosphors
   and blends containing them and light sources and displays incorporating
Optical imaging devices
   (cathodoluminescent; quantum-splitting fluoride-based phosphors and
   blends containing them and light sources and displays incorporating them)
Fluorescent lamps
   (quantum-splitting fluoride-based phosphors and blends containing them and
   light sources and displays incorporating them)
7429-91-6, Dysprosium, uses 7440-19-9, Samarium, uses
                                                          7440-27-9,
Terbium, uses
               7440-30-4, Thulium, uses 7440-53-1, Europium, uses
7440-60-0, Holmium, uses
RL: DEV (Device component use); MOA (Modifier or additive use); USES
   (activator; quantum-splitting fluoride-based phosphors and blends
   containing them and light sources and displays incorporating them)
7439-93-2, Lithium, uses
                         7440-03-1, Niobium, uses 7440-10-0,
Praseodymium, uses 7440-22-4, Silver, uses 7440-23-5, Sodium, uses
7440-31-5, Tin, uses 7440-33-7, Tungsten, uses
                                                  7440-36-0, Antimony,
      7440-50-8, Copper, uses 7440-52-0, Erbium, uses 7440-55-3,
Gallium, uses 7440-56-4, Germanium, uses 7440-58-6, Hafnium, uses 7440-62-2, Vanadium, uses 7440-66-6, Zinc, uses 7440-67-7, Zircon
                                                   7440-67-7, Zirconium,
      7440-69-9, Bismuth, uses
                                 7440-74-6, Indium, uses
                                                            18923-26-7,
Cerium 3+, uses 22541-20-4, Terbium 3+, uses
RL: DEV (Device component use); MOA (Modifier or additive use); USES
   (alkali metal gadolinium fluoride phosphors containing; quantum-splitting
   fluoride-based phosphors and blends containing them and light sources and
   displays incorporating them)
176027-02-4
RL: DEV (Device component use); USES (Uses)
   (phosphor blends containing alkali metal gadolinium fluoride phosphors and
   antimony- and europium- and manganese-activated; quantum-splitting
   fluoride-based phosphors and blends containing them and light sources and
   displays incorporating them)
13778-59-1, Lanthanum phosphate
                                 55070-88-7, Aluminum cerium magnesium
```

(phosphor blends containing alkali metal gadolinium fluoride phosphors and

cerium- and terbium-activated; quantum-splitting fluoride-based phosphors and blends containing them and light sources and displays

AΒ

ST

ΙT

ΙT

ΙT

ΙT

ΤТ

ΤТ

ΙT

them)

Phosphors

(Uses)

uses

(Uses)

oxide (All1CeMgO19)

RL: DEV (Device component use); USES (Uses)

incorporating them) ΙT 1314-36-9, Yttria, uses 12064-62-9, Gadolinium oxide (Gd2O3) 76125-60-5, Strontium aluminate (Sr4Al14025) 97358-83-3, Barium aluminate (BaAl8013) 106070-24-0, Aluminum gadolinium yttrium borate 144920-98-9, Strontium borate metaphosphate oxide (Al3(Gd,Y)(BO3)4) (Sr2(B03)0.32(P03)1.6800.68) 675819-83-7 841303-44-4 869368-09-2 869368-11-6 869368-12-7 869368-14-9 875485-03-3 RL: DEV (Device component use); USES (Uses) (phosphor blends containing alkali metal gadolinium fluoride phosphors and europium-activated; quantum-splitting fluoride-based phosphors and blends containing them and light sources and displays incorporating them) 12159-91-0, Germanium magnesium fluoride oxide (GeMg4F05.5) ΙT RL: DEV (Device component use); USES (Uses) (phosphor blends containing alkali metal gadolinium fluoride phosphors and manganese-activated; quantum-splitting fluoride-based phosphors and blends containing them and light sources and displays incorporating them) ΙT 7439-96-5, Manganese, uses 7440-45-1, Cerium, uses 16397-91-4, Manganese 2+, uses 16910-54-6, Europium 2+, uses 19768-33-3, Manganese 4+, uses 22541-18-0, Europium 3+, uses 23713-48-6, Antimony 3+, uses RL: DEV (Device component use); MOA (Modifier or additive use); USES (Uses) (phosphor blends containing alkali metal gadolinium fluoride phosphors and phosphors activated with; quantum-splitting fluoride-based phosphors and blends containing them and light sources and displays incorporating 106804-21-1, Magnesium strontium phosphate ((Mg,Sr)3(PO4)2) ΙT RL: DEV (Device component use); USES (Uses) (phosphor blends containing alkali metal gadolinium fluoride phosphors and tin-activated; quantum-splitting fluoride-based phosphors and blends containing them and light sources and displays incorporating them) 13573-11-0, Magnesium tungstate (MgWO4) 36989-78-3 104663-37-8, ΤТ Gadolinium magnesium borate (GdMgB5010) 473908-53-1 473908-57-5 RL: DEV (Device component use); USES (Uses) (phosphor blends containing alkali metal gadolinium fluoride phosphors and; quantum-splitting fluoride-based phosphors and blends containing them and light sources and displays incorporating them) 26916-87-0P, Lithium gadolinium fluoride (LiGdF4) 38670-03-0P, Potassium gadolinium fluoride (KGdF4) 896506-18-6P RL: DEV (Device component use); SPN (Synthetic preparation); PREP (Preparation); USES (Uses) (quantum-splitting fluoride-based phosphors and blends containing them and light sources and displays incorporating them) RE.CNT 16 THERE ARE 16 CITED REFERENCES AVAILABLE FOR THIS RECORD (1) Anon; JP 07315992 1995 CA (2) Anon; JP 1143669 1999 (3) Anon; WO 02097859 2002 CA (4) Feldman, C; Journal of Luminescence 2001, V92, P245 (5) Feldmann; US 6600260 B2 2003 CA (6) Karbowiak; Jor alloy and compound 2004, V380(1), P321 (7) Khaidukov, N; Optical Materials 2002, V19, P365 CA (8) Kondo, H; Journal of Luminescence 2004, V108, P59 CA (9) Liu, B; Journal of Luminescence 2003, V101, P155 CA (10) Oskam; US 20020185961 A1 2002 (11) Oskam; US 20020190645 A1 2002 (12) Oskam, K; Journal of Alloys and Compounds 2000, V300, P421 (13) Wegh, R; Journal of Luminescence 1999, V82, P93 CA

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(16) You; Jour Lumine 2004, V110(3), P95 CA

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L10 ANSWER 3 OF 6 CA COPYRIGHT 2008 ACS on STN
AN
    144:97410 CA Full-text
    Entered STN: 26 Jan 2006
ED
ΤI
    LED-based edge lit illumination system
ΙN
    Jacob, Cherian; Chen, Chen-Lun Hsing; Radkov, Emil; Srivastava, Alok Mani;
    Setlur, Anant Achyut; Comanzo, Holly Ann; Shiang, Joseph
PA
    Gelcore, LLC, USA
SO
    U.S. Pat. Appl. Publ., 10 pp.
    CODEN: USXXCO
DT
    Patent
    English
LA
INCL 257098000
    73-11 (Optical, Electron, and Mass Spectroscopy and Other Related
    Properties)
FAN.CNT 1
    PATENT NO.
                      KIND DATE
                                        APPLICATION NO.
                                                              DATE
                      ____
                             _____
                                         ______
    US 2006001036
                       A1
                                        US 2004-884205
                              20060105
                                                              20040702
PRAI US 2004-884205
                              20040702
CLASS
PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES
_____
US 2006001036 INCL 257098000
               IPCI H01L0033-00 [I,A]
                IPCR H01L0033-00 [I,A]; H01L0033-00 [I,C]
               NCL 257/098.000
                ECLA S02B; S02B; S02B; S02B
     An edge lit illumination system providing backlight utilizing a luminescent
AΒ
     impregnated lightquide is described comprising an LED radiation source
     providing a first radiation and a lightguide optically coupled to the LED
     radiation source including a luminescent material embedded or coated on an
     output surface of the lightguide designed to absorb the first radiation, and
     emit one or more radiations, where the illumination system may further include
     addnl. optical components such as reflective layers, for directing radiation
     striking the back surfaces of the light guide back into the lightguide, as
     well as diffusion layers, UV reflectors, and polarizers. A lightquide for use
     with an LED light source in an edge lit lighting assembly is also described
     comprising an optically transmissive monolith having an input surface, a back
     surface, and an output surface; and a radiation conversion material capable of
     absorbing a first radiation at a first wavelength and emitting a second
     radiation at a second wavelength; wherein the radiation conversion material is
     at least one of dispersed in the lightquide, coated on the output surface of
     the lightquide, and dispersed in a film on the output and/or back surface of
     the light quide.
    LED light illumination source light conversion phosphor waveguide
ST
ΙT
    Electroluminescent devices
    Light sources
      Optical waveguides
       (LED-based edge lit illumination system using phosphor doped light
    12525-03-0, Calcium lanthanum sulfide (CaLa2S4) 12535-38-5, Strontium
ΤT
    yttrium sulfide (SrY2S4)
                            20775-37-5, Barium magnesium silicate
    (Ba3MgSi2O8) 76125-60-5, Aluminum strontium oxide (Al14Sr4O25)
    82992-94-7, Calcium strontium sulfide ((Ca,Sr)S)
                                                    97358-83-3, Aluminum
    barium oxide (Al8BaO13) 99533-22-9, Calcium magnesium chloride silicate
    (Ca8MgCl2(SiO4)4) 173525-28-5 223757-06-0, Gadolinium sodium borate
    oxide (Gd2Na2(BO3)20) 473908-53-1 473908-57-5
                                                    675819-82-6, Aluminum
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barium calcium strontium oxide (Al2(Ba,Ca,Sr)O4) 675819-83-7

675819-84-8, Barium calcium strontium silicate ((Ba,Ca,Sr)2(SiO4))

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675819-85-9
                 675819-86-0
                                675819-88-2
                                             675819-91-7 675819-92-8
     683211-40-7, Barium calcium silicon strontium nitride ((Ba,Ca,Sr)2Si5N8)
     841303-43-3 841303-44-4 841303-47-7, Lutetium tungsten yttrium oxide
     ((Lu,Y)2WO6)
                  841303-50-2
                                841303-51-3
                                               864429-55-0
                                                            872458-25-8
     872458-26-9
     RL: DEV (Device component use); USES (Uses)
        (LED-based edge lit illumination system using phosphor doped light
     7439-96-5, Manganese, uses 7439-98-7, Molybdenum, uses
     Potassium, uses 7440-27-9, Terbium, uses 7440-36-0, Antimony, uses
     7440-45-1, Cerium, uses 7440-53-1, Europium, uses
     RL: DEV (Device component use); MOA (Modifier or additive use); USES
     (Uses)
        (LED-based edge lit illumination system using phosphor doped light
       quide)
     872458-24-7, Calcium strontium phosphate ((Ca,Sr)10(PO4)6)
     RL: DEV (Device component use); USES (Uses)
        (mixture with boron oxide; LED-based edge lit illumination system using
        phosphor doped light guide)
     1303-86-2, Boron oxide (B2O3), uses
     RL: DEV (Device component use); USES (Uses)
        (mixture with calcium strontium phosphate; LED-based edge lit
        illumination system using phosphor doped light guide)
     1309-48-4, Magnesium oxide (MgO), uses
     RL: DEV (Device component use); USES (Uses)
        (mixture with magnesium fluoride and germanium oxide; LED-based edge lit
        illumination system using phosphor doped light guide)
     7783-40-6, Magnesium fluoride (MgF2)
     RL: DEV (Device component use); USES (Uses)
        (mixture with magnesium oxide and germanium oxide; LED-based edge lit
        illumination system using phosphor doped light guide)
     1310-53-8, Germanium oxide (GeO2), uses
     RL: DEV (Device component use); USES (Uses)
        (mixture with magnesium oxide and magnesium fluoride; LED-based edge lit
        illumination system using phosphor doped light guide)
     1314-11-0, Strontium oxide (SrO), uses
     RL: DEV (Device component use); USES (Uses)
        (mixture with phosphorus oxide and boron oxide; LED-based edge lit
        illumination system using phosphor doped light guide)
     76461-00-2, Strontium silicate (Sr2Si3O8)
     RL: DEV (Device component use); USES (Uses)
        (mixture with strontium chloride; LED-based edge lit illumination system
        using phosphor doped light quide)
     1314-56-3, Phosphorus oxide (P205), uses
     RL: DEV (Device component use); USES (Uses)
        (mixture with strontium oxide and boron oxide; LED-based edge lit
        illumination system using phosphor doped light guide)
     10476-85-4, Strontium chloride (SrCl2)
     RL: DEV (Device component use); USES (Uses)
        (mixture with strontium silicate; LED-based edge lit illumination system
       using phosphor doped light guide)
L10 ANSWER 4 OF 6 CA COPYRIGHT 2008 ACS on STN
    143:34642 CA Full-text
    Entered STN: 30 Jun 2005
    Nonlinear Optical Crystal YxLayScz(BO3)4 (x + y + z = 4)
    Ye, Ning; Stone-Sundberg, Jennifer L.; Hruschka, Michael A.; Aka, Gerard;
     Kong, Wei; Keszler, Douglas A.
     Department of Chemistry, Oregon State University, Corvallis, OR,
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97331-4003, USA

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SO
     Chemistry of Materials (2005), 17(10), 2687-2692
     CODEN: CMATEX; ISSN: 0897-4756
PΒ
    American Chemical Society
DT
    Journal
     English
LA
CC
     73-10 (Optical, Electron, and Mass Spectroscopy and Other Related
     Properties)
     Section cross-reference(s): 75
AΒ
     The new nonlinear optical crystal YxLayScz(BO3)4 (x + y + z = 4) was
     discovered. Phase boundaries were established in the determination of the x,
     y, z composition parameters that define the existence region of the trigonal
     huntite-type structure. From single-crystal x-ray diffraction measurements,
     the member Y0.57La0.72Sc2.71(BO3)4 crystallized in space group R32 with a
     9.774(1) and c 7.944(3) Å. Large single crystals were grown by a high-
     temperature solution method. The high-energy optical absorption edge for
     polished pieces is at a wavelength <200 nm. Sellmeier equations for the
     dispersion in the refractive indexes were determined from curve fitting of
     data obtained by the method of min. deviation. From modeling and optical
     measurements on powders, the nonlinear optical coefficient dll is 1.4 pm/V.
ST
     nonlinear optical crystal lanthanum scandium yttrium borate
ΙT
    Crystal structure
       Optical absorption edge
       Optical dispersion
     Refractive index
        (of lanthanum scandium yttrium borate nonlinear optical
       material)
ΙT
    Nonlinear optical materials
        (preparation and crystal structure of lanthanum scandium yttrium borate)
ΙT
     853030-12-3, Lanthanum scandium yttrium borate
     (La0.77Sc2.95Y0.28(BO3)4) 853030-13-4, Lanthanum scandium
     yttrium borate (La0.76Sc2.92Y0.32(BO3)4) 853030-14-5, Lanthanum
     scandium yttrium borate (La0.8Sc2.82Y0.38(BO3)4) 853030-15-6,
     Lanthanum scandium yttrium borate (La0.73Sc2.85Y0.42(BO3)4)
     853030-16-7, Lanthanum scandium yttrium borate
     (La0.75Sc2.78Y0.47(BO3)4)
     RL: PRP (Properties)
        (preparation and crystal lattice parameters of nonlinear optical
        material)
TТ
     853030-11-2, Lanthanum scandium yttrium borate
     (La0.72Sc2.71Y0.57(BO3)4)
     RL: PRP (Properties)
        (preparation and crystal structure of nonlinear optical material)
ΙT
     648431-00-9, Lanthanum scandium yttrium borate
     (La0.7Sc3Y0.3(BO3)4)
     RL: PRP (Properties)
        (preparation and m.p. of nonlinear optical material)
RE.CNT 19
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    1993
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